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PPLICATION NO		FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO	
10/091,784		03/06/2002	Wayne M. Barnes	60019630-0038	9712	
26263	7590	02/14/2006		EXAM	EXAMINER	
SONNEN	SCHEIN	I NATH & ROSEN	CHUNDURU, SURYAPRABHA			
P.O. BOX WACKER		STATION, SEARS TO	OWER	ART UNIT	PAPER NUMBER	
CHICAGO, IL 60606-1080				1637	- · · · · · · · · · · · · · · · · · · ·	

DATE MAILED: 02/14/2006

Please find below and/or attached an Office communication concerning this application or proceeding.

Advisory Action Before the Filing of an Appeal Brief

Application No.	Applicant(s)
10/091,784	BARNES ET AL.
Examiner	Art Unit
Suryaprabha Chunduru	1637

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	Suryaprabha Chunduru	1637	
The MAILING DATE of this communication app	ears on the cover sheet with the	correspondence add	lress
THE REPLY FILED 23 January 2006 FAILS TO PLACE THIS	APPLICATION IN CONDITION FO	OR ALLOWANCE.	
 The reply was filed after a final rejection, but prior to or this application, applicant must timely file one of the foll places the application in condition for allowance; (2) a N (3) a Request for Continued Examination (RCE) in comfollowing time periods: 	on the same day as filing a Notice lowing replies: (1) an amendment, Notice of Appeal (with appeal fee) i	of Appeal. To avoid al affidavit, or other evid n compliance with 37 (ence, which CFR 41.31; or
a) The period for reply expiresmonths from the mailing b) The period for reply expires on: (1) the mailing date of this Ac event, however, will the statutory period for reply expire later t	dvisory Action, or (2) the date set forth in t	the final rejection, whichev	er is later. In no
Examiner Note: If box 1 is checked, check either box (a) or (b MONTHS OF THE FINAL REJECTION. See MPEP 706.07). ONLY CHECK BOX (b) WHEN THE		D WITHIN TWO
Extensions of time may be obtained under 37 CFR 1.136(a). The date of	n which the petition under 37 CFR 1.136	(a) and the appropriate ext	ension fee have
been filed is the date for purposes of determining the period of extension CFR 1.17(a) is calculated from: (1) the expiration date of the shortened sabove, if checked. Any reply received by the Office later than three monte earned patent term adjustment. See 37 CFR 1.704(b). NOTICE OF APPEAL	and the corresponding amount of the fee statutory period for reply originally set in the	 The appropriate extension of (2) 	on fee under 37) as set forth in (b)
 The Notice of Appeal was filed on <u>23 January 2006</u>. A the date of filing the Notice of Appeal (37 CFR 41.37(a) appeal. Since a Notice of Appeal has been filed, any re), or any extension thereof (37 CFF	R 41.37(e)), to avoid d	ismissal of the
AMENDMENTS			
3. The proposed amendment(s) filed after a final rejection (a) They raise new issues that would require further (b) They raise the issue of new matter (see NOTE be	consideration and/or search (see N		because
(c) They are not deemed to place the application in b appeal; and/or		reducing or simplifying	g the issues for
(d) They present additional claims without canceling NOTE: (See 37 CFR 1.116 and 41.33(a)		rejected claims.	
4. The amendments are not in compliance with 37 CFR 1	•	Compliant Amendmen	t (PTOL-324)
5. Applicant's reply has overcome the following rejection		zomphana / mionamon	. (
Newly proposed or amended claim(s) would be the non-allowable claim(s).		e, timely filed amendr	nent canceling
7. For purposes of appeal, the proposed amendment(s): a how the new or amended claims would be rejected is proposed. The status of the claim(s) is (or will be) as follows:		will be entered and an	explanation of
Claim(s) allowed: <u>none</u> . Claim(s) objected to: <u>none</u> .			
Claim(s) rejected: <u>1-7.</u>			
Claim(s) withdrawn from consideration: <u>8-13</u> . <u>AFFIDAVIT OR OTHER EVIDENCE</u>			
 The affidavit or other evidence filed after a final action, because applicant failed to provide a showing of good and was not earlier presented. See 37 CFR 1.116(e). 	but before or on the date of filing a and sufficient reasons why the affid	Notice of Appeal will avit or other evidence	not be entered is necessary
9. The affidavit or other evidence filed after the date of filing entered because the affidavit or other evidence failed to showing a good and sufficient reasons why it is necess	overcome <u>all</u> rejections under app	eal and/or appellant fa	ails to provide a
10. ☐ The affidavit or other evidence is entered. An explanal REQUEST FOR RECONSIDERATION/OTHER	tion of the status of the claims after	r entry is below or atta	ched.
11. The request for reconsideration has been considered See Continuation Sheet.			ance because:
12. Note the attached Information Disclosure Statement(s	s). (PTO/SB/08 or PTO-1449) Pape	er No(s)	n
13. Other:	B	nobha Chumd IRYAPRABAA C 7ENT EXAM	LM 2/8/06
	Sc PA	RYAPRABAAC JENT EXAM	HUNDURU

Continuation of 11, does NOT place the application in condition for allowance because: Applicants' arguments are found unpersuasive. With regard to the rejection under 35 USC102 (b) as being anticipated by Lin et al. Applicants' arguments are fully considered and found unpersuasive. Applicants argue that Lin et al. does not teach MgHPO4 in col. 7, line 44-49 as suggested by the examiner and did not teach a source of phosphate ions and a source of magnesium ions form a precipitate at a temperature below 34 C and argue that the solution taught by Lin et al. is used for lypolization purposes rather than for use as a precipitate and thus Lin et al. teaches away from the instant invention. Applicants' arguments are fully considered and found unpersuasive. First, Examiner clearly mentioned that the "wherein" clause and the limitations followed by wherein clause are inherent property of a kit ,that comprises a source of phosphate and a source of magnesium ions. Thus the arguments based on wherein clause, forming a precipitate at a temperature below 34 C are unpersuasive because the kit taught by Lin et al. does comprise a source of phosphate ions (Na2HPO4) and a source of magnesium ions (MgCl2). As noted in MPEP 2112.01 "Products of identical chemical composition can not have mutually exclusive properties. "A chemical composition and its properties are inseparable. Therefore, if the prior art teaches the identical chemical structure, the properties applicant discloses and/or claims are necessarily present. In re Spada, 911 F.2d 705, 709, 15 USPQ2d 1655, 1658(Fed. Cir. 1990). It is also noted that according to the MPEP 2112.02 -The discovery of a new use for an old structure based on unknown properties of the structure might be patentable to the discoverer as a process of using. In re Hack, 245 F.2d 246, 248, 114 USPQ 161, 163 (CCPA 1957). However, when the claim recites using an old composition or structure and the "use" is directed to a result or property of that composition or structure, then the claim is anticipated. In re May, 574 F.2d 1082, 1090, 197 USPQ 601, 607 (CCPA 1978). Thus the use of old composition to show new property have no patentable value. Thus the phrase "wherein combining the source of magnesium ions and the source of phosphate ions in accordance with the instructions supplied in the kit forms a precipitate at a temperature below 34 C" is an inherent property of the kit comprising a source of magnesium ions and a source of phosphate ions and instructions for use of the kit. Further, instructions for using the source of phosphate and magnesium ions, is drawn to an intened use which is not given any patentable weiight because a recitation of the intended use of the claimed invention must result in a structural difference between the claimed invention and the prior art in order to patentably distinguish the claimed invention from the prior art. If the prior art structure is capable of performing the intended use, then it meets the claim. In a claim drawn to a process of making, the intended use must result in a manipulative difference as compared to the prior art. See In re Casey, 152 USPQ 235 (CCPA 1967) and In re Otto, 136 USPQ 458, 459 (CCPA 1963). Applicants' assertions regarding MgHPO4 is incorrect because Examiner did not discuss such component in the rejection. With regard to the assertions on Lin et al. teaches away from the instant invention, Examiner notes that according to MPEP 2145, "A prior art reference that "teaches away" from the claimed invention is a significant factor to be considered in determining obviousness; however, "the nature of the teaching is highly relevant and must be weighed in substance. A known or obvious composition does not become patentable simply because it has been described as somewhat inferior to some other product for the same use." In re Gurley, 27 F.3d 551, 554, 31 USPQ2d 1130, 1132 (Fed. Cir. 1994), a teaching away, is a significant factor to be considered as "teaching in"., Examiner notes that the solution taught by Lin et al. is not for the purpose of lypolization because in col. 7, line 44-49, Lin et al. teach that the kit components can be packed either as a lipolized form or in a solution form. Therefore the rejection is maintained herein.

With regard to the rejection under 35 USC 103(a) as being unpatentable over Bloch et al. in view of Stanley, Applicants' arguments are fully considered and found unpersuasive. Applicants argue that the combination of Bloch et al. in view of Stanely did not suggest or teach the instant claims because Stanely did not teach magnesium as a suitable lypotropic salts, i.e., a salt which has the property of promoting precipitation (salting out)and argues that Stanely teaches away from the use of magensium by reciting only monovalently charged cations and does not teach lypotropic salts precipitate at temperatures below 34 C as required by claim 1. Applicants' arguments are fully considered and found unpersuasive. The rejection under 35 USC 103(a) is based on alternatively taking the limitations followed by "wherein clause" into consideration. Stanely taught some suitable lypotropic salts., and it does not mean that other salts having lypotropic property are eliminated from the list mentioned by Stanley. Further Stanely et al. does teach salting-out at temperature below 34 C as discussed in the rejection. With regard to Applicants' assertions that there is no suggestion or teaching to combine the references, examiner notes that obviousness can only be established by combining or modifying the teachings of the prior art to produce the claimed invention where there is some teaching, suggestion, or motivation to do so found either in the references themselves or in the knowledge generally available to one of ordinary skill in the art. See In re Fine, 837 F.2d 1071, 5 USPQ2d 1596 (Fed. Cir. 1988) and In re Jones, 958 F.2d 347, 21 USPQ2d 1941 (Fed. Cir.1992). In the instant context, as discused in the rejection, Bloch et al. teach Kit composition including source of magnesium and silent regarding the source of phosphate ions and Stanely teaches phosphoric acid as a source of phosphate ions and provides motivation to use phosphoric acid as a donor of phosphate ions. Thus as discussed in the rejection one skilled in the art would be motivated to combine the kit as disclosed by Bloch et al. in a manner taught by Stanley because Stanley explicitly taught the use of a source of phosphate ions, and use of such ions in promoting precipitation (salting-out) of certain types of high molecular weight species (at temperatures ranging from 20-25 C), such as proteins in the aqueous solution and thereby enhancing the attachment of molecular species (oligonucleotides) to the water-soluble intermediate reagent (nucleic acid target) (see col. 7, line 18-43).

With regard to the Applicants' assertion drawn to the Stanely reference teaching away from the instant invention as discussed above MPEP 2145 states, "A prior art reference that "teaches away" from the claimed invention is a significant factor to be considered in determining obviousness; however, "the nature of the teaching is highly relevant and must be weighed in substance. A known or obvious composition does not become patentable simply because it has been described as somewhat inferior to some other product for the same use." In re Gurley, 27 F.3d 551, 554, 31 USPQ2d 1130, 1132 (Fed. Cir. 1994), a teaching away, is a significant factor to be considered as "teaching in". In the instant context the teachings of Stanely are based on the use of a source of phosphate ions and Bloch reference teach all the limitations including a source of magnesium and it is obvious that the combination of phosphate ions to the kit components comprising a source of magnesium would result in a precipitate (slting out) as discussed in the rejection. Therfore the rejection is maintained herein.

claims (dependent on claim 1) obvious. As discussed above, the combiantion of Bloch et al. in view of Stanely does make the claim 1 obvious and the combination of Bloch et al. in view of Stanely further in view of Barnes does make the instant claims obvious for the use of thermostable DNA polymerase as taught by Barnes et al. Therefore the rejection is maintained herein.